Constraining the rate of oceanic deoxygenation leading up to a Cretaceous Oceanic Anoxic Event (OAE-2: ∼94 Ma)

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The rates of marine deoxygenation leading to Cretaceous Oceanic Anoxic Events are poorly recognized and constrained. If increases in primary productivity are the primary driver of these episodes, progressive oxygen loss from global waters should predate enhanced carbon burial in underlying sediments—the diagnostic Oceanic Anoxic Event relic. Thallium isotope analysis of organic-rich black shales from Demerara Rise across Oceanic Anoxic Event 2 reveals evidence of expanded sediment-water interface deoxygenation ∼43 ± 11 thousand years before the globally recognized carbon cycle perturbation. This evidence for rapid oxygen loss leading to an extreme ancient climatic event has timely implications for the modern ocean, which is already experiencing large-scale deoxygenation.

INTRODUCTION

Anthropogenic forcing of the ocean-atmosphere system necessitates an understanding of Oceanic Anoxic Events (OAEs) and the associated time scales and mechanisms of initiation to predict potential future marine deoxygenation (1). The cause of OAE-2, which occurred at the Cenomanian–Turonian boundary (~94 Ma (million years ago)), has been attributed to various combinations of increased surface temperatures (2), sea level rise (3), nutrient trapping (4), and pulses of magmatic activity (5–7)—all of which are likely to increase primary productivity (7). In the modern predominantly oxic ocean, primary productivity in the euphotic zone produces excess organic carbon that subsequently sinks and is remineralized to CO2 by heterotrophic bacteria, in the process consuming free oxygen. Thus, increases in primary productivity are accompanied by increases in oxygen consumption and should lead to a progressive expansion in anoxic portions of the ocean (8). Recent advances place coarse temporal constraints on select OAE-2 trigger mechanisms (9–11) and provide evidence for this gradual marine deoxygenation (12–16), but the timing of the transition from widespread ocean oxygenation before OAE-2 to the peak of ocean deoxygenation remains effectively unconstrained. Uncertainty surrounding the exact duration of OAE-2—and thus also surrounding the timing of the what is going on before and during the event—adds further complications, with recent estimates ranging from ∼450 to 700 ky (thousand years) (17–19).

The massive burial of organic carbon during OAE-2 [perhaps 1.4 to 1.6 times that of the steady-state modern ocean (14, 20)], which caused the observed carbon isotope excursion (2, 5), must be linked to changes in redox conditions even when considering caveats from sedimentation rate changes (21, 22). Manganese (Mn) oxides are ubiquitous in sediments deposited below an oxygenated water column but are absent under anoxic or euxinic conditions, because quantitative chemical reduction of insoluble Mn (IV) to more soluble Mn (II) and Mn (III) occurs when oxygen is essentially removed from the water in contact with the sediment (23). Attempts to directly track perturbations in both local and global Mn oxide burial have been proven difficult because of numerous factors, but primarily because Mn budgets in ancient marine sediments are often perturbed by postdepositional processes (24).

Thallium (Tl) strongly adsorbs to Mn oxides during their precipitation with a large positive isotope fractionation factor, rendering seawater isotopically light (25, 26). Global Mn oxide precipitation is the primary control of the marine Tl isotope composition on time scales shorter than ~5 million years (27), and the modern marine residence time of Tl is ~18.5 ky (28). Therefore, reconstructions of the Tl isotope composition of seawater for short-term events, such as OAEs, should provide evidence of perturbations in Mn oxide precipitation. Recent work shows that, because Tl is quantitatively removed from seawater under euxinic conditions, sediments underlying these environments faithfully record the Tl isotope composition of overlying waters (29). As long as the euxinic setting is well connected to the open ocean (for example, Caraiaco Basin, Venezuela), sediments in these locations record the global seawater value. Thus, ancient euxinic sediments are an archive capable of tracking changes in the global Mn oxide burial flux in cases where the overlying water column was well connected to open ocean seawater.

Here, we present Tl isotope compositions of organic-rich euxinic black shales before, during, and after OAE-2 that were recovered from ODP (Ocean Drilling Program) Site 1258 located on Demerara Rise in the Atlantic Ocean (table S1) (30). The stratigraphy and temporal framework of cores from Site 1258 sampled for this study are well constrained (see the Supplementary Materials for more details). The macroscopic view of the entire section has been shown to record local euxinic conditions based on Fe speciation (15) while, at the same time, being relatively well connected with the open ocean because of its paleogeographic location (fig. S1) (30). These observations show that samples from ODP Site 1258 are ideal for reconstructing changes in Mn oxide burial before, during, and after OAE-2 using Tl isotopes.

RESULTS

Before the event, Tl isotope compositions are slightly heavier than modern seawater values (r205Tl = −6) (31, 32), averaging r205Tl ∼ −4.5 and believed to represent connection to well-oxygenated global ocean (Fig. 1A; implications for this value are discussed in the Supplementary Materials). Most markedly, a distinct positive shift to r205Tl ∼ −2.5 occurs before the onset of OAE-2 and is maintained until slightly before the

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termination of the event, where Tl isotopes start recovering toward lighter values of $\varepsilon^{205}\text{Tl} \sim -3$. Low-resolution black shale analysis before and during OAE-2 from a second basin (Furlo, Marche-Umbria, Italy) reveals nearly identical values to Site 1258 both before and during enhanced organic carbon burial (fig. S2), suggesting that these two records faithfully record the global ocean Tl isotope composition with little or no significant basinal heterogeneity or restriction affecting Tl isotopes in these two locations.

**DISCUSSION**

We conclude that the positive Tl isotope excursion reflects diminished global Tl scavenging by Mn oxides due to an expansion of areas in the ocean with minimal to no oxygen at the sediment-water interface. The Tl isotope excursion commences significantly before the carbon isotope excursion at Site 1258 (Fig. 1B) (33). Using a linear sedimentation rate extrapolated from the OAE, and assuming that the duration of OAE-2 was 600 ky, we calculate a time lag of $\sim 43 \pm 11$ ky between the initiation of the Tl and C isotope excursion (see the Supplementary Materials).

Therefore, our data imply that deoxygenation of the sediment-water interface (as indicated by Tl isotopes) was initiated $\sim 43 \pm 11$ ky before expansion of global euxinia, as indicated by C and S isotope systematics (14). The 600-ky duration for OAE-2 used here is also the value used in recent attempts to constrain the timing of possible OAE-2 trigger mechanisms (10, 11), guided by the GSSP (Global Boundary Stratotype Section and Point) time scale and estimates from multiple sites (17, 18). If the duration of OAE-2 was actually shorter or longer than the 600 ky assumed here, then our estimate of $\sim 43 \pm 11$ ky is too long or too short, respectively.

The decrease in Mn oxide burial before OAE-2 can be modeled using a previously used isotope mass balance model (27). Although euxinic sediments accumulate significant Tl concentrations and, therefore, may constitute a significant Tl sink when marine euxinia expands, anoxic sediments deposited underneath oxygen minimum zones (OMZs) appear to have essentially little to no Tl enrichment (34). Therefore, we can treat the initial deoxygenation before OAE-2 where marine euxinia was yet to expand in terms of the same inputs and outputs that govern the modern marine Tl isotope reservoir. The shift from $\varepsilon^{205}\text{Tl} = -4.5$ to $-2.5$ before OAE-2 can be accounted for if Mn oxide output fluxes decreased by $\sim 40$ to $80\%$, depending on the Tl isotope fractionation factor between seawater and Mn oxides (see Fig. 2 and the Supplementary Materials for further details).

To our knowledge, this is the first study to track global perturbations in Mn oxide burial leading to a major climatic event. We can use these changes in Mn oxide burial as a direct measure of the seafloor area affected by anoxia. For this study, we refer to sediments being deposited beneath a sediment-water interface with oxygen concentrations close to 0, at least long enough to prevent Mn oxide burial, as anoxic. Thus, we can convert the $\sim 40$ to $80\%$ change in Mn oxide burial flux into the area of anoxic seafloor by assuming that the areal extent of Mn oxide deposition was the primary control of Mn oxide burial fluxes over short time scales. Therefore, our model suggests that over a period of $\sim 43 \pm 11$ ky, the area of anoxic seafloor expanded by 40 to 80\%, or at a rate of $\sim 2.2 \times 10^3$ to $7.5 \times 10^3$ km$^2$/year, substituting for previously oxygenated portions of the sediment-water interface (see the Supplementary Materials). If these oxygen depletions primarily manifested themselves via expansion of areas similar to modern OMZs, then more...
than the entire continental shelf and upper slope (~16% of the modern seafloor area) but did not recover significantly even after organic carbon burial had largely diminished to pre-OAE levels. This post-OAE decoupling of the two isotope systems could imply that although overall organic matter burial declined, productivity was still sufficiently high to maintain the consumption of any oxygen that might have been subducted into the deep ocean through ocean circulation. Such a conclusion is supported by the fact that Os isotope values do not return to pre-event radiogenic values in most of the studied sites (10, 11), which may indicate that significant LIP activity continued even after the recovery of C isotopes. In this case, the nutrient supply that fueled the initial increase in primary productivity could have remained relatively high and possibly maintained the expanded state of global anoxia. Alternatively, it has been hypothesized that the termination of increased carbon burial during OAE-2 was due to the drawdown of bioessential trace metals from seawater (15), which might have inhibited some parts of the carbon cycle and, therefore, reduced carbon burial even in the face of enhanced nutrient supply from LIP activity. Additionally, after the main OAE-2 event, a second positive C isotope excursion (between 421.19 and 420.98 mcd (Fig. 1)) and coeval drop in V concentrations are accompanied by a sudden positive Tl isotope excursion. These geochemical shifts are identical to those associated with the main OAE event, but the magnitudes are much smaller. Although this conclusion is speculative, this feature may imply a shorter-lived spike of deoxygenation and enhanced carbon burial that occurred shortly after OAE-2.

On the basis of our interpreted sequence of events leading up to OAE-2, the general marine response to increased nutrient supply, global warming, and sea level rise is expansion of oceanic anoxia without simultaneous expansion of ocean euxinia. Increased ocean deoxygenation is already apparent in the modern ocean, because marine O$_2$ has decreased by 2% over roughly the last half century (41), and recent models predict a continued loss of 0.5 to 3.5% over the next half century (42). Should anthropogenically induced oxygen loss occur at similar rates as in the period leading up to OAE-2, then the current area of seafloor hypoxia would double in the next 102 to 344 years, equivalent to an increase of $2.2 \times 10^4$ to $7.5 \times 10^4$ km$^2$ of seafloor every decade—an easily detectable value. Localized oxygen loss is already apparent in the modern ocean (1, 41, 42), and the ability to observe more widespread perturbation seems realistic under currently projected carbon emissions. Without positive human intervention, ancient OAE studies are destined to become uncomfortably applicable in the not-so-distant future.

**MATERIALS AND METHODS**

**Sample digestion**

Previously powdered samples were digested using a technique modified from previous literature (34). Samples were first treated with 2 M HNO$_3$...
to separate authigenic and detrital components, and the supernatant was isolated from the silicate residue. This partial dissolution was immediately followed by 1.5 hours of digestion at ~300°C and 10 bars in a high-pressure asher to help dissolve organic compounds. Following dissolution of the silicate residue in 1:1 concentrated HNO₃ and HF, both fractions were completely dissolved in 1 M HCl in preparation for ion exchange chromatography.

**TI purification**
Thallium was isolated from a sample matrix using previously described anion exchange chromatography techniques (31, 44). However, unlike previous studies, only one column was used during TI purification. Testing of standards and samples showed that there was no difference in concentration or isotope composition between samples that had been processed through one or two columns (table S2).

**Sample analysis**
The TI isotope compositions were determined at the Woods Hole Oceanographic Institution (WHOI) Plasma Mass Spectrometry Facility using a Thermo Finnigan Neptune multiple collector inductively coupled plasma mass spectrometer. Isotopic compositions are reported relative to the National Institute of Standards and Technology (NIST) 997 TI metal in epsilon notation

\[ \epsilon^{205}\text{TI} = \left( \frac{^{205}/203\text{TI}_{\text{sample}}}{^{205}/203\text{TI}_{\text{NIST 997}}} - 1 \right) \times 10^4 \]

Previously described techniques that used both external normalization to NIST SRM (Standard Reference Materials) 981 Pb and standard-sample bracketing were applied for mass bias correction (31, 44). Because of the quantitative yields of TI from the column chemistry procedure, TI concentrations were determined by monitoring the \(^{205}\text{TI}\) signal in concert with the quantitative yields of TI from the column chemistry procedure. Testing of standards and samples showed that there was no difference in concentration or isotope composition between samples that had been processed through one or two columns (table S2).

**Supplementary materials**
Supplementary material for this article is available at http://advances.sciencemag.org/cgi/content/full/3/8/e1701020/DC1

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