Future methane, hydroxyl, and their uncertainties: key climate and emission parameters for future predictions

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Received: 1 August 2012 – Accepted: 1 August 2012 – Published: 20 August 2012
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Published by Copernicus Publications on behalf of the European Geosciences Union.

Abstract

Accurate prediction of future methane abundances following a climate scenario requires understanding the lifetime changes driven by anthropogenic emissions, meteorological factors, and chemistry-climate feedbacks. Uncertainty in any of these influences or the underlying processes implies uncertainty in future abundance and radiative forcing. We simulate methane lifetime in multiple models over the period 1997–2009, adding sensitivity tests to determine key variables that drive the year-to-year variability. Across three atmospheric chemistry and transport models – UCI CTM, GEOS-Chem, and Oslo CTM3 – we find that temperature, water vapor, ozone column, biomass burning and lightning NOₓ are the dominant sources of interannual changes in methane lifetime. We also evaluate the model responses to forcings that have impacts on decadal time scales, such as methane feedback, and anthropogenic NOₓ emissions. In general, these different CTMs show similar sensitivities to the driving variables. We construct a parametric model that reproduces most of the interannual variability of each CTM and use it to predict methane lifetime from 1980 through 2100 following a specified emissions and climate scenario (RCP 8.5). The parametric model propagates uncertainties through all steps and provides a foundation for predicting methane abundances in any climate scenario. Our sensitivity tests also enable a new estimate of the methane global warming potential (GWP), accounting for stratospheric ozone effects, including those mediated by water vapor. We estimate the 100-yr GWP to be 32.

1 Introduction

Rising atmospheric concentrations of greenhouse gases are the main cause of current and future climate change (Intergovernmental Panel on Climate change IPCC, 2007). Uncertainty in mapping an emission scenario onto future abundance of greenhouse gases (GHGs) thus translates almost directly into uncertainty in our ability to
project climate change and its impact on nature and society. To date, IPCC has generally adopted a single trajectory for the growth of greenhouse gases in each of several different socio-economic scenarios, thus neglecting uncertainty in those future abundances. For methane, the second most important anthropogenic GHG, these trajectories are based on simple parametric formulas for methane lifetime. In the IPCC Third Assessment Report (TAR), 4 parameters accounted for the change in tropospheric OH, the largest atmospheric methane sink, due to anthropogenic emissions of CO, nitrogen oxides (NOx), and volatile organic compounds (VOCs) and the negative feedback between methane abundance and tropospheric OH (Prather et al., 2001). Other sinks, which include oxidation in the stratosphere, oxidation by tropospheric chlorine, and uptake into soil, were assessed but assumed not to change during the 21st century projections. For the upcoming IPCC 5th Assessment Report (AR5) the Representative Concentration Pathway (RCP) scenarios adopt methane trajectories calculated in the MAGICC model, which augments the TAR parametric formula with a temperature term (Meinshausen et al., 2011a).

On small spatial scales, OH concentrations and methane oxidation depend on temperature, pressure, sun elevation, clouds, UV attenuation by stratospheric ozone, and local concentrations of water vapor, ozone, CH4, CO, NOx, VOCs, and aerosols (e.g. Duncan et al., 2000; Olson et al., 2006). Integrated globally and annually, some of these influences are small, but numerous studies have found that temperature, circulation, water vapor, stratospheric ozone, clouds and natural and anthropogenic emissions are important (Dentener et al., 2003; Fiore et al., 2006; Hess and Mahowald, 2009; Lelieveld and Crutzen, 1994; Stevenson et al., 2005; Voulgarakis et al., 2010). Uncertainties in these factors and in the present-day methane budget mean that each socioeconomic emission scenario could produce a range of future methane abundances (Prather et al., 2012).

Global climate model (GCM) simulations with atmospheric chemistry provide another method for predicting future methane and other chemically reactive GHGs. An ensemble of such models can provide a range of future methane abundances for a single scenario (e.g. Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP) Lamarque et al., 2012), spanning some, but likely not all, future uncertainties. This approach is computationally expensive, however, which restricts the number of socioeconomic scenarios and ensemble members that can be explored.

In this work we develop a new parametric model for global methane lifetime that accounts for climate-chemistry interactions that were neglected in previous approaches. We derive the parametric factors from perturbation tests in a suite of 3 chemical transport models (CTMs), since CTMs with detailed tropospheric chemistry provide the best mechanistic representation of methane loss from tropospheric OH. We focus on the tropospheric OH sink because other methane sinks are smaller and their intrinsic variability has a smaller impact on the total methane lifetime. The parametric model accounts for uncertainty in atmospheric chemistry based on the range of perturbation responses across the CTMs. The perturbation tests also enable a new estimate of the ozone contribution to methane radiative forcing and global warming potential. We evaluate the parametric model against 13-yr CTM simulations of methane lifetime, and against observed variability in tropospheric OH, as measured by the decay of methyl chloroform. Finally, we use this parametric model with uncertainties to make new projections of methane and its uncertainties through 2100.

2 Model descriptions

We diagnose methane lifetime due to tropospheric OH, $\tau_{CH_4-OH}$, from multi-year simulations in 3 different CTMs: University of Oslo CTM3, University of California, Irvine (UCI) CTM, and GEOS-Chem. All of these models are driven by assimilated meteorological data and configured to use the same emissions from anthropogenic, biogenic and biomass burning sources. We use year-specific meteorology spanning 1997–2009 for each model, except GEOS-Chem simulations with GEOS-5 meteorology, which are only 2004–2009 (see below). Sections 2.1–2.4 summarize unique features of each model and describe the emissions.
Monthly chemistry diagnostics from each model enable us to calculate $\tau_{CH_4 \times OH}$, defined as the total atmospheric CH$_4$ burden divided by its loss through reactions with tropospheric OH. All 3 models use fixed methane abundances (1760 ppb for UCI CTM and CTM3, 1775 ppb for GEOS-Chem), so variations in $\tau_{CH_4 \times OH}$ are due solely to changes in the OH sink. Different tropopause definitions in the models have minimal effect on $\tau_{CH_4 \times OH}$ since CH$_4$ oxidation between 200 hPa and the tropopause is 1.5% of tropospheric methane loss, or less. We calculate the total methane lifetime, $\tau_{CH_4}$, using $\tau_{CH_4 \times OH}$ values from this work and recently estimated lifetimes for other methane sinks: tropospheric chlorine (200 yr), stratosphere (120 yr), and soil (150 yr) (Prather et al., 2012).

2.1 Oslo CTM3

Oslo CTM3 is a stratospheric and tropospheric CTM, recently described by Søvde et al. (2012). Transport is driven by pieced-forecast meteorology from the European Center for Medium-range Weather Forecasting (ECMWF) Integrated Forecast System (cycle 36r1, http://www.ecmwf.int/research/lfsdocs/CY36r1/index.html). The original T359 ($\sim 0.55^\circ \times 0.55^\circ$) horizontal resolution and 60 layer vertical resolution of the forecast model is degraded to T42 ($\sim 2.8^\circ \times 2.8^\circ$) resolution, while preserving the 3 h temporal resolution for all meteorological fields. Advection uses the second-order moments scheme (Prather, 1986; Prather et al., 2008) and convection follows Tiedtke (1989).

The CTM3 chemical mechanism includes a full stratospheric chemical mechanism in addition to tropospheric reactions. The tropospheric module contains 105 reactions and 51 gas-phase species, including sulfate, nitrate, and sea-salt aerosols. Nitrate aerosols affect the gas-phase chemistry through HNO$_3$ uptake, which is a sink for reactive nitrogen through subsequent wet scavenging. Photolysis rates required in the chemistry mechanism are calculated online using the Fast-JX method (Neu et al., 2007), with cloud distributions from ECMWF meteorology. CTM3 shares the same chemical mechanism and some other physical process algorithms with the older CTM2, which has been extensively used for studies of present and future tropospheric composition (Dalsoren et al., 2010; Hoor et al., 2009; Isaksen et al., 2005).

2.2 UCI CTM

The UCI CTM is a tropospheric CTM, using the same meteorology, transport algorithms, and Fast-JX photolysis as CTM3. Like CTM3, the UCI CTM uses T42 horizontal resolution, but the vertical resolution in the boundary layer is reduced, so there are 57 layers total. Tropospheric chemistry of the major gas-phase species involved in HO$_x$, NO$_x$, O$_3$, and VOC reactions is simulated with the ASAD package (Carver et al., 1997), with updates to the mechanism and kinetics (Tang and Prather, 2010). This mechanism includes 84 reactions involving 33 species, making it simpler than the CTM3 chemical mechanism. Simplified stratospheric O$_3$ chemistry is simulated with Linoz (version 2 Hsu and Prather, 2009). Aerosol effects on photolysis and chemistry are neglected, which increases OH and biases $\tau_{CH_4 \times OH}$ high by about 10% (Bian et al., 2003; Mcintyre and Evans, 2010; Martin et al., 2003).

2.3 GEOS-Chem

GEOS-Chem is a tropospheric CTM, driven by assimilated meteorological data from the NASA Goddard Earth Observing System (GEOS-5) or MERRA reanalysis (Rienecker et al., 2011, 2008). Both GEOS-5 and MERRA are produced from closely related assimilation systems, using the same spatial resolution of 0.55° x 0.66° and 72 vertical layers. Most GEOS-Chem results here, including all sensitivity simulations, are based on GEOS-5 meteorology, which has been degraded to 2° x 2.5° and 47 layers for the CTM. GEOS-5 data are available only after 2004, however, so we also simulate 1997–2009 using MERRA meteorology at 4° x 5° and 47 layers. Temporal resolution in GEOS-5 (MERRA) is 6 h (3 h) for most meteorological quantities and 3 h (1 h) for surface quantities and mixing depth.
The tropospheric chemistry mechanism in GEOS-Chem consists of 104 species and 236 chemical reactions that simulate aerosols in addition to the HO$_4$-NO$_x$-VOC-ozone system, and has recently been updated by Mao et al. (2010). Photolysis rates are calculated with the Fast-JX method, using aerosol optical depths that are simulated internally, and ozone columns from the TOMS and SBUV satellites (until 2008) or GEOS-5 assimilation of satellite data (after 2008). For purposes of stratosphere-troposphere exchange, stratospheric ozone is simulated with Linoz.

### 2.4 Emissions

Emissions used in this work are representative of 1997–2010, but do not resolve trends or interannual variability in anthropogenic or biogenic emissions. To the extent possible, we use identical emissions across all models. Anthropogenic, biogenic, and biomass burning emissions of NO$_x$, CO, and isoprene are fully consistent in all models. Some differences in VOC emissions arise because of the different lumping schemes used in the various chemical mechanisms and because some VOC species are not simulated in all models. Lightning NO$_x$ emissions also differ between models because they are calculated from underlying meteorology, as described below.

Table 1 summarizes emissions of key species. We use the RCP inventory for anthropogenic emissions for year 2000, repeating in each simulated year (Lamarque et al., 2010; van Vuuren et al., 2011). This inventory provides monthly gridded emissions for NO$_x$, CO and speciated VOCs from 11 emission activities. Aviation and shipping emissions change each month, while other anthropogenic emission activities are constant throughout the year. Biomass burning emissions are specified for each year and month by the GFED inventory (version 3 van der Werf et al., 2010). We use this instead of the climatological biomass burning emissions provided in the RCP inventory because fires are a major cause of year-to-year variability in tropospheric OH. Biogenic emissions of isoprene, CO, and other VOCs are from a MEGAN climatology for the 2000s decade (Guenther et al., 2006). GEOS-Chem includes additional oceanic emissions of acetone (13 Tg yr$^{-1}$ Jacob et al., 2002) and acetaldehyde (57 Tg yr$^{-1}$ Millet et al., 2010), which are not included in other models. All emission data are provided at 0.5° × 0.5° resolution.

Lightning NO$_x$ emissions (L-NO$_x$) are calculated with similar methods in all 3 CTMs, with UCI CTM and CTM3 using identical algorithms. In all models, these emissions are derived from cloud-top heights in the underlying meteorology (Price and Rind, 1994) and scaled to match satellite-observed lightning flash rates (Christian et al., 2003). In the UCI CTM and CTM3, 2 scale factors are calculated to match observed multi-year mean flash rates over land and ocean. In GEOS-Chem scale factors are calculated for every grid column and month (Sauvage et al., 2007). Within the convective column, L-NO$_x$ is distributed vertically based on observed NO$_x$ distributions (Ott et al., 2010). Sevde et al. (2012) provide a full description of lightning emissions in UCI CTM and CTM3, and Murray et al. (2012) do the same for GEOS-Chem. L-NO$_x$ averages 6 Tg(N) yr$^{-1}$ in GEOS-Chem and 5 Tg(N) yr$^{-1}$ in UCI CTM and CTM3.

### 3 Recent (1997–2009) variability of CH$_4$ lifetime

Figure 1 shows $\tau$$_{CH_4-OH}$ for 1997–2009, as simulated by the 3 CTMs. The tropospheric OH lifetimes range from 8.5 to 10.1 yr. The longest of these lifetimes (GEOS-Chem) is consistent with the constraint provided by methyl chloroform observations, 11.2 ± 1.3 yr (Prather et al., 2012), but all are within the range of contemporary tropospheric chemistry models (e.g. 9.5 ± 1.1 yr from ACCMIP Naik et al., 2012).

These simulations show similar variability of $\tau$$_{CH_4-OH}$ in all CTMs. Common features include a sharp dip in 1998 and peak in 2000, coincident with a strong El Niño and La Niña, smaller peaks in 2004 and 2008, and general decline after 2005. These features appear robust against the various choices of chemical mechanism, meteorology, and resolution used in these CTMs. In independent work, the ECHAM model also simulates the same features, using different emissions and chemistry (Montzka et al., 2011).
3.1 Methane lifetime sensitivity to chemistry-climate factors

Having identified robust variations in $\tau_{\text{CH}_4 \times \text{OH}}$ across multiple CTMs, we examine their causes with explicit perturbation tests. In these tests, we perturb a single climate or emission variable, simulate 3 or more years, discard the first year as spinup, and analyze the difference from the unperturbed simulation in the remaining years. Perturbations are applied to 1997–1999 for Oslo CTM3 and the UCI CTM, and to 2004–2006 for GEOS-Chem with GEOS-5 meteorology. The sensitivity, $\alpha$, of $\tau_{\text{CH}_4 \times \text{OH}}$ to a climate or emission variable, $F$, is always defined as $\alpha = d \ln(\tau_{\text{CH}_4 \times \text{OH}})/d \ln(F)$. As such, $\alpha$ can be interpreted as the percent change in $\tau_{\text{CH}_4 \times \text{OH}}$ resulting from a 1% increase in $F$.

Table 2 reports sensitivities for the evaluated climate and emission variables. These variables include most of those identified in the literature as important influences on tropospheric OH and $\tau_{\text{CH}_4 \times \text{OH}}$: temperature, water vapor, ozone column, convective fluxes, cloud optical depth, biomass burning emissions, and NO$_x$ emissions. Perturbation magnitudes are chosen to be similar to the interannual variability or decadal trend of each variable (exact magnitudes in Table S1).

Only variables with large sensitivity, large interannual changes, or both can explain the year-to-year $\tau_{\text{CH}_4 \times \text{OH}}$ variations identified in Fig. 1. Figure 2 shows the interannual changes of 5 key variables for 1997–2009. Water vapor, having about 3% variation and $\tau_{\text{CH}_4 \times \text{OH}}$ sensitivity near –0.3, could account for about 1% interannual variability in $\tau_{\text{CH}_4 \times \text{OH}}$. Temperature, ozone column, L-NO$_x$, and biomass burning also have sufficient sensitivity and variability to account for about 1% variation in $\tau_{\text{CH}_4 \times \text{OH}}$ over the 1997–2009 period. These 5 climate and emission variables that we identify as important influences on $\tau_{\text{CH}_4 \times \text{OH}}$ have been recognized previously, but their sensitivities have not been quantified in a comparable way (e.g., Dentener et al., 2003; Fiore et al., 2006; Hess and Mahowald, 2009; Stevenson et al., 2005).

Convective fluxes and cloud optical depths for water and ice clouds, as diagnosed in ECMWF meteorology, vary annually by 2% and have small sensitivity, so these factors have very little impact on $\tau_{\text{CH}_4 \times \text{OH}}$. Due to the small impact in the UCI CTM, these perturbation tests are not repeated in the other CTMs. Our results are consistent with the known decrease in mass-weighted global OH concentrations due to clouds (Voulgarakis et al., 2010) because mass-weighted averaging emphasizes below-cloud OH concentrations and we find compensating increases in methane loss above clouds.

In addition, past analyses of convective fluxes have found both positive and negative influences on $\tau_{\text{CH}_4 \times \text{OH}}$ depending on the convection scheme and perturbation used (Lawrence and Salzmann, 2008).

Methane abundance and anthropogenic NO$_x$ emissions increased over the 2000–2010 decade by 1% and 2%, respectively, and vary smoothly between years (Dlugokencky et al., 2011; Granier et al., 2011). Therefore, these factors have little impact on $\tau_{\text{CH}_4 \times \text{OH}}$ Variability during the 13-yr CTM simulation, but are important on multi-decadal time scales and longer.

Many of the sensitivity terms in Table 2 – specifically, water vapor, biomass burning, CH$_4$ abundance, and anthropogenic land NO$_x$ – are consistent among the CTMs and with past estimates (Fiore et al., 2006; Hoor et al., 2009; Myhre et al., 2011; Prather et al., 2001), suggesting a good understanding of how these variables impact tropospheric methane loss. Adopted values for each sensitivity (Table 2, right column), which are used in the parametric model described below, reflect the consistency among models. For biomass burning, the agreement between models masks large changes in sensitivity between years, shown in Fig. 3 for the UCI CTM. The sensitivity is highly correlated with total biomass burning emissions, and the CO/NO ratio in those emissions, both of which suppress tropospheric OH (Duncan et al., 2003; Voulgarakis et al., 2010) and peak during El Niño years due to tropical peat fires. Future climate may be more El Niño-like due to GHG warming (Yamaguchi and Noda, 2006), so, despite the CTM consensus on present-day biomass burning sensitivity, we adopt a broad uncertainty range for future sensitivity.

Other sensitivities, chiefly air temperature and ship NO$_x$, differ by 50% or more across the models. These differences are understandable, however, as consequences of modeling assumptions. For ship NO$_x$, CTM3 and UCI CTM are nearly 3 times more
sensitive than GEOS-Chem. In the UCI CTM and CTM3, ship NOx is emitted as NO, diluted into the grid volume, and the subsequent production of O3 and HNO3 are calculated by the grid-resolved chemistry. Instantaneous dilution overestimates the NOx lifetime and O3 production from ships (Chen et al., 2005), however. To compensate, GEOS-Chem instantaneously converts all ship NOx emissions to O3 and HNO3, following observed production ratios. As a result, GEOS-Chem underestimates the large-scale impact of shipping, since, in reality, 20–50% of NOx remains after 5 h following emission (Vinken et al., 2011). Although previous estimates of ship NOx are close to the high values in this work (Hoer et al., 2009; Myhre et al., 2011), the actual atmospheric sensitivity to ship NOx, likely lies somewhere between the GEOS-Chem and UCI CTM results.

Ship NOx emissions also explain the divergence of GEOS-Chem and the UCI CTM in their temperature sensitivities. Over land, both models predict similar reduction in CH4×OH in response to warming. Over the oceans, however, GEOS-Chem predicts longer CH4×OH at higher temperatures while the UCI CTM predicts the opposite. In the presence of ship NOx, in the UCI CTM, higher temperatures increase both the production and loss of O3, with net excess production; OH rises in turn. In GEOS-Chem, by contrast, higher temperatures increase O3 destruction over the ocean with less opportunity for enhanced production; OH thus decreases over oceans.

The sensitivity of CH4×OH to methane abundance is closely related to the methane feedback factor, r, which is the ratio of methane perturbation lifetime to total budget lifetime (Prather et al., 2001). Our multi-model mean sensitivity, 0.31 ± 0.04, is similar to past estimates (Fiore et al., 2009; Prather et al., 2001), but we derive a smaller feedback factor r = 1.34 ± 0.06 than has been recommended by IPCC (r = 1.4 Prather et al., 2001) because we use updated estimates of methane lifetime (Prather et al., 2012). Reducing the feedback factor, which was already suggested by Fiore et al. (2009), lowers the methane radiative forcing and global warming potential, as discussed in Sect. 3.5.

### 3.2 A parametric model for CH4×OH

The sensitivity parameters in Table 2, together with the time series of corresponding climate and emission variables in Fig. 2, enable us to build a parametric model for methane lifetime representing each CTM. We combine terms linearly, so that CH4×OH is approximated by

\[
\ln(\tau_{CH4×OH}(t)) = \ln(\tau_{CH4×OH}) + \sum \alpha_i \Delta \ln(F_i(t)),
\]

where \(F_i(t)\) is the time series of forcing variable \(i\) and \(\tau_{CH4×OH}\) is the mean lifetime in the CTM. Figure 1 shows the parametric model reconstruction of each CTM, alongside the actual calculated \(\tau_{CH4×OH}\). We find that 5 variables – temperature, water vapor, column ozone, biomass burning emissions, and L-NOx emissions – explain 90% of the interannual variation in \(\tau_{CH4×OH}\) in the UCI CTM and GEOS-Chem over the simulated period 1997–2009. Even though the GEOS-Chem sensitivity parameters were derived from 2° × 2.5° simulations driven by GEOS-5, the 5-parameter model performs equally well compared to the 4° × 5° GEOS-Chem simulation driven by MERRA. The sensitivity parameters are thus robust across changes in model resolution and meteorology. For Oslo CTM3 the 5-parameter model explains 50% of \(\tau_{CH4×OH}\) variability overall, rising to 75% outside the 1997–1998 ENSO. A higher temperature sensitivity, similar to the UCI CTM, in the parametric model also increases the explained variance for CTM3 to 80%.

The atmospheric chemistry of tropospheric OH and methane involves nonlinear chemistry that could, in principle, undermine the additivity of terms in Eq. (1). We test the linearity of the system with a final perturbation test in the UCI CTM in which all 5 factors are perturbed simultaneously. The resulting change in \(\tau_{CH4×OH}\) differs by about 1 part in 10 from the linear addition of factors.

The CTM simulations in this work make several assumptions to simplify the perturbation analysis and enable comparisons between CTMs, but these could alter \(\tau_{CH4×OH}\). In particular, the simulations neglect variability in biogenic VOC emissions (Guenther...
et al., 2006), trends in anthropogenic emissions and their location, and trends in atmospheric methane. We compare our GEOS-Chem/MERRA simulation to one that includes all these processes, and other minor model updates, and find correlations of 98 % for $\tau_{\text{CH}_4 \times \text{OH}}$ (M. Mu, personal communication, 2012). Thus, the neglected processes make small to interannual variability of $\tau_{\text{CH}_4 \times \text{OH}}$ and do not degrade the parametric model performance.

3.3 Methyl chloroform comparison

Two global measurement networks have recorded the growth and decline of atmospheric methyl chloroform (MCF) since the 1970s (ALE/GAGE/AGAGE Prinn et al., 2005), with expanded coverage since the 1990s (NOAA Montzka et al., 2000). Like methane, atmospheric MCF is oxidized mainly by tropospheric OH, with small additional sinks in the stratosphere, oceans, and soil (Volk et al., 1997; Wang et al., 2008; Wennberg et al., 2004). Because MCF has no natural sources and the anthropogenic production is well known (McCulloch et al., 1999), MCF provides the best available constraint on global OH levels and methane lifetime. The analysis here uses observations since 1998, when anthropogenic MCF emissions became small compared to atmospheric oxidation of the residual atmospheric burden. Consequently, MCF atmospheric lifetimes can be inferred from observed decay rates without detailed accounting for emissions and transport (Montzka et al., 2011).

For each network, we calculate decay rates of MCF from monthly average concentrations provided by each network (NOAA: ftp://ftp.cmdl.noaa.gov/hats/solvents/CH3CCl3/flasks/GCMS/CH3CCL3_GCMS_flask.txt, last access: 6 August 2012; AGAGE: http://agage.eas.gatech.edu/data_archive/agage/gc-md/monthly/, last access: 4 April 2012). For site $i$ and month $t$ the observed decay rate (yr$^{-1}$) is

$$k_{i,t} = -\ln(c_{i,t+(\delta t)}/c_{i,t-(\delta t)}),$$

where $c_{i,t}$ is the concentration at site $i$ in month $t$. The global MCF decay rate is the average of $k_{i,t}$ across sites within a network. We calculate uncertainty in the global decay rate as the 16th–84th percentile range (i.e. $\pm 1\sigma$) of $k_{i,t}$ across sites within a network. No filling is used for months with missing data. (See Supplement for additional method details.) Over 1998–2007, the global MCF decay rates from the 2 networks differ by less than 1 % ($0.1811 \text{ a}^{-1}$ for NOAA, 0.1796 $\text{a}^{-1}$ for AGAGE). This analysis, however, focuses on anomalies in the global decay rate, relative to each network’s own mean. Because the anomalies are attributed solely to tropospheric OH loss (see below) and for comparison to $\tau_{\text{CH}_4 \times \text{OH}}$, the decay anomalies are divided by $r = 0.87$ to account for the tropospheric OH fraction of total MCF loss (Prather et al., 2012).

Figure 4 compares the interannual variability of simulated $\tau_{\text{CH}_4 \times \text{OH}}$ in the CTMs against the MCF decay rate. While the CTMs are consistently within the observational uncertainty for both observation networks, the year-to-year changes in the models generally do not correlate with the MCF data. In addition, simulated $\tau_{\text{CH}_4 \times \text{OH}}$ in all CTMs varies less than the MCF constraint (1 % vs. 2.3 % for $\text{MCF}$/mean). Residual anthropogenic or ocean emissions could account for some MCF decay rate anomalies, but only if these emissions change abruptly from year to year. Emission anomalies of about 4 Gg yr$^{-1}$ would be required to cause the observed decay rate swings during 2002–2004. Meanwhile, total anthropogenic and ocean emissions for those years are estimated to be 6 and 4 Gg yr$^{-1}$, respectively, and decreasing smoothly (Montzka et al., 2011; Prinn et al., 2005; Wennberg et al., 2004). Therefore, abrupt emission changes might explain part, perhaps half, of the decay anomalies, but cannot account for the full discrepancy between simulated $\tau_{\text{CH}_4 \times \text{OH}}$ and observations.

Collocated measurement sites in the NOAA and AGAGE networks provide an alternative means to evaluate possible errors in decay rates. At all 4 collocated sites (Cape Grim, American Samoa, Trinidad Head, and Mace Head) we find differences between the networks as large as 2 % in the monthly means. (See Fig. S2) The differences exceed the standard error in monthly mean and persist for several consecutive months; thus they are likely not caused by synoptic variability and differences in sampling frequency. Because the biases change over time, they lead to differences of up to 4 % in MCF decay rates at a single site. As can be seen in Fig. 2, both networks find similar
magnitude of OH variability, but they differ in sign and magnitude of the anomaly at many times. Given that differences in observed MCF decay rates between the two networks are as large as their difference from CTM $\tau_{\text{CH}_4 \times \text{OH}}$ anomalies, we conclude that better understanding of the systematic differences between the observation networks is required before using them as a constraint on $\tau_{\text{CH}_4 \times \text{OH}}$ and OH interannual variability.

### 3.4 Methane global warming potential

Global Warming Potentials (GWP) are useful for comparing the radiative forcing (RF) caused by emissions of various GHGs having different absorbances and atmospheric lifetimes. The methane GWP customarily accounts for the direct RF from the emitted gas, as well as indirect RF caused by methane-induced increases in ozone, stratospheric water vapor, and feedback on the methane lifetime (Forster et al., 2007). Here we evaluate the methane GWP implied by the perturbation experiments. Radiative forcing of methane and ozone are calculated for the control simulation and a simulation with 5% more methane, using the University of Oslo radiative transfer model (Myhre et al., 2011). In addition, we test the effect of methane-induced water vapor on stratospheric ozone, with an additional Oslo CTM3 simulation in which stratospheric water vapor was increased to maintain equilibrium with the CH$_4$ perturbation. To our knowledge, this indirect, H$_2$O-mediated effect on ozone has not been included in prior assessments of methane GWP.

Table 5 summarizes ozone changes and RF results for all simulations, normalized to 1 ppb CH$_4$ perturbations. Tropospheric ozone changes in GEOS-Chem and the UCI CTM (2.9 and 4.0 DU ppm(CH$_4$)$^{-1}$, respectively) are within the range of previous multi-model studies (Fry et al., 2012; Holmes et al., 2011), Oslo CTM3, however, exhibits larger tropospheric changes (5.0 DU ppm(CH$_4$)$^{-1}$), likely due to the effects of stratospheric chemistry on the upper troposphere. Stratospheric ozone changes (10.3 DU ppm(CH$_4$)$^{-1}$) are twice as large as the tropospheric changes, but still small compared to the total stratospheric column. We find that stratospheric water vapor produced by oxidation of methane causes small decreases in stratospheric ozone ($\sim$4.3 DU ppm(CH$_4$)$^{-1}$).

Ozone generally has greater radiative forcing efficiency in the troposphere than in the stratosphere (Forster and Shine, 1997), so tropospheric ozone changes tend to dominate the ozone RF components. In our 3 models, tropospheric ozone RF is 30–50% of the direct methane RF, and up to 65% after including stratospheric ozone mediated by methane and water. Previous IPCC assessments have assumed 25% for purposes of calculating GWP (Forster et al., 2007; Shine et al., 1995), similar to a recent estimate of 21% based on tropospheric changes alone (Fry et al., 2012). Methane perturbation data from the TAR (3.67 DU(O$_3$) ppm(CH$_4$)$^{-1}$) (Prather et al., 2001), however, suggest that tropospheric ozone RF is about 40% of the methane RF (154 mW m$^{-2}$ ppm(CH$_4$)$^{-1}$, assuming efficiency of 42 mW m$^{-2}$ DU(O$_3$)$^{-1}$) (Ramawamy et al., 2001).

Accounting for both direct and indirect effects, the methane RF efficiency, $F_e$, is 618 mW m$^{-2}$ ppm(CH$_4$)$^{-1}$ in steady-state. A 1 Tg pulse emission of methane raises the atmospheric abundance by $\delta = 0.364$ ppb, which decays at a rate $f_{\text{CH}_4}$, where $f = 1.33$ is the methane feedback on its lifetime. We use $\tau_{\text{CH}_4} = 9.14$ yr (Prather et al., 2012). Neglecting delays between emission time and stratospheric impacts, the 100-yr absolute GWP is $\delta f_{\text{CH}_4} F_e = 2.75$ mW yr m$^{-2}$, compared to 0.087 mW yr m$^{-2}$ for CO$_2$.

Thus, the methane GWP$_{100}$ is 31.6. Our result is higher than several previous reports, generally near 25 (Forster et al., 2007; Fry et al., 2012), mainly because we include stratospheric ozone effects, but also because the updated and longer methane lifetime used here (Prather et al., 2012). IPCC TAR recommended $f = 1.4$ (Prather et al., 2001), which would imply an even larger GWP, but since $f$ depends on $\tau_{\text{CH}_4}$ the two must be chosen consistently. Uncertainty in the GWP is difficult to assess without further modeling and analysis of stratospheric impacts, but it is likely $\pm$20% or larger.
4 Historical (1980–2005) changes in CH₄ lifetime

Having established the ability of Eq. (1) to reconstruct τₐCH₄×OH over 1997–2009 in CTMs, we now use it to extrapolate methane lifetime over several decades for which the CTMs have not been run. We begin with the historical period 1980–2005, during which time the key atmospheric forcing variables have been relatively well observed by satellites and ground stations.

In addition to the 5 climate and emission variables identified in Sect. 3.2 as important influences on interannual variability, we include CH₄ abundance and anthropogenic NOₓ emissions from land, ships, and aircraft for the historical reconstruction. We also include sensitivity to anthropogenic CO and VOC emissions, based on the IPCC TAR parameters, but without uncertainties (Prather et al., 2001). In total, the expanded parametric model includes 11 parameters and variables. For the sensitivity parameters, αᵢ in Eq. (1), we adopt values from the average and spread of sensitivities in the 3 CTMs (Table 2, last column).

Table 3 summarizes the data sources for historical climate and emission variables in the expanded parametric model. NASA MERRA reanalysis provides temperature and water vapor data (Bosilovich et al., 2011) and satellite observations provide ozone column data (Stolarski and Frith, 2006). As with the 5-parameter model, these are averaged over the latitudes, 40° S to 40° N, that are important for CH₄ oxidation. Historical CH₄ abundance and anthropogenic and biomass burning emissions follow CMIP5 recommendations (Lamarque et al., 2010; Meinshausen et al., 2011b). Global annual lightning flash rates have varied by up to 20 % since 1998, but multi-decadal trends are not apparent (Murray et al., 2012), so we assume no change since 1980, with 10 % Gaussian uncertainty in the trend.

Figure 5 shows the historical changes in τₐCH₄×OH reconstructed from Eq. (1), together with the contribution from each of the climate and emission variables. To account for uncertainties in parameters and the lightning forcing, we generate 10⁵ monte carlo realizations of Eq. (1), allowing all parameters to vary independently. The resulting uncertainty in the τₐCH₄×OH reconstruction, measured as standard deviation across the realizations, reflects uncertainty in the parameters, αᵢ, but not uncertainty in emissions, ozone observation, or meteorological assimilation. (See Fig. S6 for uncertainties in τₐCH₄×OH in each component.)


Overall, the parametric model simulates a decrease in τₐCH₄×OH since 1980, which has also been found in numerous CTM and GCM studies (Dentener et al., 2003; Duncan et al., 2000; Hess and Mahowald, 2009; Karlssottr and Isaksen, 2000; Naik et al., 2012; Stevenson et al., 2005). This is an improvement over previous parametric approaches, which are shown in Fig. 5, that produce zero or positive trends over the same period (Meinshausen et al., 2011a; Prather et al., 2001).

Figure 6 identifies the contribution of each variable to the total change in τₐCH₄×OH. Rising atmospheric methane has the largest influence on τₐCH₄×OH, but the positive methane feedback effect (4 %) is more than compensated by negative climate and emission terms. Temperature and water vapor, which have increased due to GHGs, decrease τₐCH₄×OH by 2 %, collectively, although the water vapor effect is about 3 times larger. Halogen-driven decreases in stratospheric ozone also shortened the lifetime about 1 %. Increases in ship and land anthropogenic NOₓ emissions both decrease τₐCH₄×OH by 1.5 %, despite the ship source having much smaller total magnitude. Lightning NOₓ could also have an important impact on τₐCH₄×OH, but the lightning trends are not known.

The total τₐCH₄×OH change from 1980–1985 to 2000–2005 is −2.3 ± 1.8 % in our model, or −0.13 % yr⁻¹ from a linear fit. Dentener et al. (2003), simulated a larger decrease, −0.2 % yr⁻¹, in the 1980s that they attributed mainly to water vapor.
Meteorological inputs may contribute to the difference, since water vapor trends are known to vary amongst reanalysis products (Trenberth et al., 2011). In addition, the shift of anthropogenic emissions to SE Asia, which alters the sensitivity of $\tau_{CH_4 \times OH}$ to emissions is not treated in the parametric model (e.g. Fuglestvedt et al., 1999; Karlsson et al., 1999). Methyl chloroform analyses generally suggest large decreases in $\tau_{CH_4 \times OH}$ during the 1980s followed by increases during the 1990s, which conflicts with the CTM results (Bousquet et al., 2005; Prinn et al., 2005). Assuming uncertainty of about 20% in methyl chloroform emissions, however, reconciles the observations with the small trends found in CTMs and in our parametric model (Krol and Lelieveld, 2003; Prinn et al., 2005).

5 Future (2010–2100) CH$_4$ and CH$_4$ lifetime

We now apply the parametric model to predict methane and methane lifetime, with their uncertainties, following a future socioeconomic scenario. We make predictions for RCP 8.5 (Riahi et al., 2007), a scenario with rapid climate warming, but these methods apply to other scenarios as well. The prediction begins with the best estimate of present-day methane budget, including natural and anthropogenic emissions, and lifetimes for all loss processes, using the method of Prather et al. (2012). The scenario specifies future anthropogenic methane emissions and we assume natural emissions could change ±20% (1σ) by 2100. We use the parametric model to predict future $\tau_{CH_4 \times OH}$ and adopt other loss rates from literature (Prather et al., 2012). For future predictions we use the same expanded set of 11 parameters as were used in the historical $\tau_{CH_4 \times OH}$ reconstruction (Table 2, last column). Table 3 lists data sources for the future climate and emission variables.

Table 4 summarizes the predicted changes in climate and emissions in RCP 8.5. In this scenario most anthropogenic emissions of ozone precursors decrease by 2100 (7–75%), although aircraft NOx emissions rise 123%. Biomass burning emissions, also specified by the scenario, decrease 35%. The parametric prediction requires 20949...
lie within our estimated uncertainties throughout the 21st century. Statistical uncertainties in methane predictions are 8 % in 2100, based on the assessed processes in the parametric model. Neglected processes – including shifting emission locations, biogenic VOC emissions, stratosphere-troposphere exchange, and aerosol interactions with photolysis and chemistry – might cause additional systematic prediction errors, but we have found that these have minor impact on present-day interannual $\tau_{\text{CH}_4 \times \text{OH}}$ variability.

The parametric model predicts $\tau_{\text{CH}_4 \times \text{OH}}$ will increase $+13.3 \pm 10.0$ % by 2100 (Fig. 6). MAGICC gives similar results ($+12.6$ %), but the IPCC TAR formula yields a larger result ($+29.6$ %), consistent with their respective historical performances in Sect. 4.

The ACCMIP model ensemble predicts $+6.2 \pm 10.2$ % for RCP 8.5 (Voulgarakis et al., 2012), which demonstrates that the simple parametric approach covers much of the range suggested by computationally intensive GCM ensemble integrations. Lightning NO$_x$ emissions likely explain most of the $\tau_{\text{CH}_4 \times \text{OH}}$ difference, since ACCMIP models calculate 24 % increase in 2100 (Voulgarakis et al., 2012). Although we do not think future lightning estimates from GCMs are robust (see above), assuming an equally large change in the parametric model would lower $\tau_{\text{CH}_4 \times \text{OH}}$ in 2100 by about 5 %, after accounting for methane feedback.

Figure 7 and Table 4 decompose the net $\tau_{\text{CH}_4 \times \text{OH}}$ changes in 2050 into components due to each climate and emission variable. Uncertainties here include possible errors in both the sensitivity and forcing variable, except for the emission terms where all uncertainty comes from the sensitivity parameter. Methane feedback is the largest influence, having an individual contribution of $+29.0 \pm 7.3$ %. NO$_x$ emission reductions over land also force $\tau_{\text{CH}_4 \times \text{OH}}$ upwards ($+12.8 \pm 0.9$ %), which is opposite to NO$_x$ influence in recent decades. Other climate and emission components are zero or negative, with water vapor having the largest effect ($-11.5 \pm 2.8$ %). Stratospheric ozone and lightning NO$_x$ contribute little to $\tau_{\text{CH}_4 \times \text{OH}}$ changes, but they make a significant contribution to the uncertainty.

6 Conclusions

Over 1997–2009, the 3 CTMs in this work exhibit common variability in methane lifetime, which is also shared by other published model studies. We quantitatively explain these features with 5 climate and emission variables – temperature, water vapor, ozone column, biomass burning emissions, and lightning NO$_x$ emissions. A parametric model built on these 5 factors reproduces 50–90 % of the variability in methane lifetime during 1997–2009. The ensemble of 3 models provides a measure of uncertainty in each parametric factor, which we use to project past and future methane and its lifetime, with uncertainties. While this approach lacks the full complexity of atmospheric chemistry that can be included through multi-decadal simulations of a CTM or GCM, the advantage is that it can be rapidly applied to many climate data sets or socioeconomic scenarios. Using the parametric model to reconstruct methane lifetime for 1980–2005, we estimate $2.3 \pm 1.8$ % decrease in $\tau_{\text{CH}_4 \times \text{OH}}$, which is the same direction of change as previous CTM studies but smaller magnitude. For the RCP 8.5 future scenario, methane abundances are larger than the CMIP5 recommendations, which are based on the MAGICC model, but the uncertainty encompasses the difference. Uncertainty in 2100 abundance is 10 % based on the processes we have assessed here. Water vapor, anthropogenic NO$_x$ emissions, and methane feedback on its OH sink are the major drivers of $\tau_{\text{CH}_4 \times \text{OH}}$ in both the historical and future simulations.

We also provide a new estimate of the indirect components of methane RF. Tropospheric ozone contributes 30–50 % of the direct methane RF, compared to 25 % that has been used in previous IPCC assessments (Forster et al., 2007). After including stratospheric chemistry effects, including those mediated by water vapor, we estimate the methane-induced ozone RF to be 50 % of the direct methane RF. Based on these data, the 100-yr methane GWP is 32.
Supplementary material related to this article is available online at: http://www.atmos-chem-phys-discuss.net/12/20931/2012/acpd-12-20931-2012-supplement.pdf.

Acknowledgements. We thank Mingquan Mu (UC Irvine) for providing additional GEOS-Chem simulations that we compare to our results. This research was supported by the NASA Modeling, Analysis, and Prediction Program (NNX09AJ47G), the Office of Science (BER) of the US Department of Energy (DE-SC0007021), and the Kavli Chair in Earth System Science.

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Table 1. Emissions.

<table>
<thead>
<tr>
<th>Source (Inventory)</th>
<th>NOx, Tg(N) yr⁻¹</th>
<th>CO, Tg yr⁻¹</th>
<th>Isoprene, Tg yr⁻¹</th>
</tr>
</thead>
<tbody>
<tr>
<td>Anthropogenic (RCP year 2000)</td>
<td>32b</td>
<td>609</td>
<td>–</td>
</tr>
<tr>
<td>Biomass burning (GFED3)</td>
<td>5.6²</td>
<td>360⁶</td>
<td>–</td>
</tr>
<tr>
<td>Biogenic (MEGAN)</td>
<td>–</td>
<td>76</td>
<td>523</td>
</tr>
<tr>
<td>Lightning</td>
<td>5d</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td><strong>Total</strong></td>
<td>42</td>
<td>1047</td>
<td>523</td>
</tr>
</tbody>
</table>

² Inventory references: RCP (Lamarque et al., 2010; van Vuuren et al., 2011), GFED3 (van der Werf et al., 2006).

³ Land, ship, and aviation components are 26, 5.4, and 0.85 Tg(N) yr⁻¹, respectively.

⁴ Average biomass burning for 1997–2009. Emissions for individual years are 3.3–6.1 Tg(N) yr⁻¹ and 263–605 Tg(N) yr⁻¹.

⁵ Average for 1997–2009 in UCI CTM and Oslo CTM3. Emissions for individual years are 4.8–5.4 Tg(N) yr⁻¹. GEOS-Chem has 5.7–6.4 Tg(N) yr⁻¹ (average 6 Tg(N) yr⁻¹) over 2004–2009.

Table 2. Sensitivity of τCH₄×OH to climate variables and emissions.

<table>
<thead>
<tr>
<th>Variable (Chemistry-climate interactions)</th>
<th>UCI CTM</th>
<th>Oslo CTM3</th>
<th>GEOS-Chem</th>
<th>Literature</th>
<th>Adopted</th>
</tr>
</thead>
<tbody>
<tr>
<td>Air temperature [°C]³</td>
<td>–3.9</td>
<td>–2.8</td>
<td>–2.2</td>
<td>–3.0 ± 0.8</td>
<td></td>
</tr>
<tr>
<td>Water vapor [g kg⁻¹]³</td>
<td>–0.32</td>
<td>–0.29</td>
<td>–0.34</td>
<td>–0.32 ± 0.03</td>
<td></td>
</tr>
<tr>
<td>Ozone column [N m⁻²]³</td>
<td>+0.66</td>
<td>+0.43</td>
<td>+0.61</td>
<td>+0.28–0.76 [7]</td>
<td></td>
</tr>
<tr>
<td>Lightning NO₂ emissions [a, b]</td>
<td>–0.14</td>
<td>–0.11</td>
<td>–0.24</td>
<td>–0.16 ± 0.06</td>
<td></td>
</tr>
<tr>
<td>Biomass burning emissions [a, b, d]</td>
<td>+0.021</td>
<td>+0.024</td>
<td>+0.017</td>
<td>+0.021 ± 0.010</td>
<td></td>
</tr>
<tr>
<td>CH₄ abundance [a, d]</td>
<td>+0.383</td>
<td>+0.307</td>
<td>+0.274</td>
<td>+0.32 [1]</td>
<td></td>
</tr>
<tr>
<td>Convective mass flux</td>
<td>–0.036</td>
<td></td>
<td></td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>Optical depth, ice clouds</td>
<td>+0.013</td>
<td></td>
<td></td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>Optical depth, water clouds</td>
<td>–0.025</td>
<td></td>
<td></td>
<td>N</td>
<td></td>
</tr>
</tbody>
</table>

Anthropogenic emissions

| Land NOₓ [a, b] | –0.15 | –0.10 | –0.16 | –0.137 [1] |
| Ship NOₓ [b] | –0.045 | –0.050 | –0.017 | –0.0412 ± 0.014 [4] |
| Aviation NOₓ [b] | –0.014 ± 0.003 [6] | | | –0.014 ± 0.003 [6] |
| CO [b] | +0.11 [1] | | | +0.11 |
| VOC [b] | | | | +0.047 [1] |

³ Sensitivities are of τCH₄×OH (τF) for each variable f. Values calculated for 1998–1999 perturbations simulations in UCI CTM and Oslo CTM3 and 2004–2005 for GEOS-Chem, except where noted below for biomass burning and CH₄ feedback. Perturbation magnitudes were chosen to be similar to interannual variability or decadal trend (see Table 3f for exact magnitudes).

⁴ Methane feedback factor, defined as the ratio of τCH₄×OH (τF) for each variable f. Values calculated for 1998–1999 perturbations simulations in UCI CTM and Oslo CTM3 and 2004–2005 for GEOS-Chem, except where noted below for biomass burning and CH₄ feedback. Perturbation magnitudes were chosen to be similar to interannual variability or decadal trend (see Table 3f for exact magnitudes).

⁵ Anthropogenic emissions include all anthropogenic emissions occurring over land, including combustion, agriculture, and waste.

⁶ Sensitivities are of τCH₄×OH (τF) for each variable f. Values calculated for 1998–1999 perturbations simulations in UCI CTM and Oslo CTM3 and 2004–2005 for GEOS-Chem, except where noted below for biomass burning and CH₄ feedback. Perturbation magnitudes were chosen to be similar to interannual variability or decadal trend (see Table 3f for exact magnitudes).

⁷ Methane feedback factor, defined as the ratio of τCH₄×OH (τF) for each variable f. Values calculated for 1998–1999 perturbations simulations in UCI CTM and Oslo CTM3 and 2004–2005 for GEOS-Chem, except where noted below for biomass burning and CH₄ feedback. Perturbation magnitudes were chosen to be similar to interannual variability or decadal trend (see Table 3f for exact magnitudes).

⁸ Anthropogenic emissions include all anthropogenic emissions occurring over land, including combustion, agriculture, and waste.

９ Anthropogenic emissions include all anthropogenic emissions occurring over land, including combustion, agriculture, and waste.

10 Sensitivities are of τCH₄×OH (τF) for each variable f. Values calculated for 1998–1999 perturbations simulations in UCI CTM and Oslo CTM3 and 2004–2005 for GEOS-Chem, except where noted below for biomass burning and CH₄ feedback. Perturbation magnitudes were chosen to be similar to interannual variability or decadal trend (see Table 3f for exact magnitudes).

¹¹ Methane feedback factor, defined as the ratio of τCH₄×OH (τF) for each variable f. Values calculated for 1998–1999 perturbations simulations in UCI CTM and Oslo CTM3 and 2004–2005 for GEOS-Chem, except where noted below for biomass burning and CH₄ feedback. Perturbation magnitudes were chosen to be similar to interannual variability or decadal trend (see Table 3f for exact magnitudes).
## Table 3. Datasets for historical and future $\tau_{\text{CH}_4 \times \text{OH}}$

<table>
<thead>
<tr>
<th></th>
<th></th>
<th></th>
<th></th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>L-NO$_x$</td>
<td>0 ± 10 % Assumed</td>
<td>+10 ± 20 % Assumed</td>
<td></td>
<td></td>
</tr>
<tr>
<td>(NO$_x$, CO, VOC)</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>CH$_4$ abundance</td>
<td>CMIP5 [4]</td>
<td>this workc</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>


b Future atmospheric temperature is calculated from sea-surface temperature (SST) in each CMIP5 model. Water vapor is then calculated from atmospheric temperature using standard vapor pressure formulas and assuming constant relative humidity. The range of SST in the CMIP5 models is propagated to uncertainty in air temperature and water vapor. See Supplement for details.

c We calculate future CH$_4$ using $\tau_{\text{CH}_4 \times \text{OH}}$ from Eq. (1) (adopted parameters from Table 2 and other inputs from this Table) and RCP 8.5 emissions of CH$_4$. Other sinks and natural emissions, plus their uncertainties, are from Prather et al. (2012). We also specify ±20 % uncertainty in natural CH$_4$ emissions in 2100. Uncertainties in all terms are propagated using the monte carlo approach of Prather et al. (2012).

## Table 4. Changes (2100–2010) in climate variables, emissions, and $\tau_{\text{CH}_4 \times \text{OH}}$ for RCP 8.5a.

<table>
<thead>
<tr>
<th>Variable</th>
<th>Variable change</th>
<th>$\tau_{\text{CH}_4 \times \text{OH}}$ change, %</th>
</tr>
</thead>
<tbody>
<tr>
<td>Air temperature, 40° S–40° N</td>
<td>+3.7 ± 0.9 K</td>
<td>–4.2 ± 1.5</td>
</tr>
<tr>
<td>Water vapor, 40° S–40° N</td>
<td>+38.5 ± 9.1 %</td>
<td>–11.5 ± 2.8</td>
</tr>
<tr>
<td>Ozone column, 40° S–40° N</td>
<td>+0.7 ± 3.0 %</td>
<td>+0.4 ± 1.7</td>
</tr>
<tr>
<td>Lightning NO$_x$ emissions</td>
<td>+10 ± 20 %</td>
<td>–1.2 ± 3.3</td>
</tr>
<tr>
<td>Biomass burning emissions</td>
<td>–34.8 %</td>
<td>+0.9 ± 0.4</td>
</tr>
<tr>
<td>CH$_4$ abundance</td>
<td>+78.5 ± 7.9 %</td>
<td>+29.0 ± 7.3</td>
</tr>
</tbody>
</table>

### Anthropogenic emissions

| Land NO$_x$                           | –75.3 %          | +12.8 ± 0.9                                   |
| Ship NO$_x$                           | –7.2 %           | +0.2 ± 0.1                                    |
| Aircraft NO$_x$                       | +123 %           | –1.7 ± 0.4                                    |
| CO                                   | –44.0 %          | –4.7                                           |
| VOC                                 | –11.1 %          | –0.5                                           |
| Total (this work)                    | +13.3 ± 10.0     |                                               |
| IPCC TAR Total (Prather et al., 2001) | +29.6            |                                               |
| MAGICC Total (Meinshausen et al., 2011a) | +12.6          |                                               |

a Variable changes from data sources in Table 3, except CH$_4$ abundance, which is calculated from the scenario CH$_4$ emissions and the time-evolving $\tau_{\text{CH}_4 \times \text{OH}}$ (see text).

$\tau_{\text{CH}_4 \times \text{OH}}$ component changes derived from the variable changes and the sensitivities in Table 2, including uncertainties in both.

b Surface to 400 hPa average.
### Table 5. Present-day, steady-state methane impact on ozone and radiative forcing.

<table>
<thead>
<tr>
<th></th>
<th>UCI CTM</th>
<th>Oslo CTM</th>
<th>GEOS-Chem</th>
<th>Literature</th>
<th>Adopted</th>
</tr>
</thead>
<tbody>
<tr>
<td>Ozone chemistry, DU(O₃) ppm(CH₄)⁻¹</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>d[O₃]/d[CH₄]</td>
<td>4.03(T)</td>
<td>4.98(T)</td>
<td>2.90(T)</td>
<td>3.5 ± 1.0(T)</td>
<td>b</td>
</tr>
<tr>
<td></td>
<td>10.32(S)</td>
<td></td>
<td></td>
<td>3.0 ± 0.8(T)</td>
<td>c</td>
</tr>
<tr>
<td>d[O₃]/d[H₂O] (from CH₄)</td>
<td>−0.40(T)</td>
<td></td>
<td></td>
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</tr>
<tr>
<td></td>
<td>−4.35(S)</td>
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<td>Radiative forcing, mW m⁻² ppm(CH₄)⁻¹</td>
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<td>CH₄ from CH₄</td>
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<td>370 ± 27d</td>
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<td></td>
<td>141(T)</td>
<td>198(T)</td>
<td>108(T)</td>
<td>126 ± 45(T)b</td>
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<td>150(T)</td>
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<tr>
<td>O₃ from CH₄ via H₂O</td>
<td>−16(T)</td>
<td>−16(T)</td>
<td>−19(S)</td>
<td></td>
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<tr>
<td>H₂O from CH₄</td>
<td>55°</td>
<td>55°</td>
<td></td>
<td>55°</td>
<td></td>
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<tr>
<td>Total</td>
<td></td>
<td></td>
<td></td>
<td>618</td>
<td></td>
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<tr>
<td>100-yr GWP</td>
<td></td>
<td></td>
<td></td>
<td>25°</td>
<td>31.6</td>
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<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>24.2 ± 4.2e</td>
<td></td>
</tr>
</tbody>
</table>

a Troposphere (T) and stratosphere (S) values given separately, wherever possible. All CTM results are for 2009.
b Review by Holmes et al. (2011).
c Fry et al. (2012).
d Forster et al. (2007).
e 15% of CH₄ direct RF (Myhre et al., 2007).
f Calculated from tropospheric O₃ change using the average RF efficiency from the other models.

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**Fig. 1.** Methane lifetime due to oxidation by tropospheric OH ($τ_{CH₄×OH}$) simulated by each CTM (solid lines) and reconstructed from the 5-parameter model (dashed lines). The parameters are temperature, water vapor, ozone column, lightning NOₓ emission, and biomass burning emission. Parameter values for each CTM are given in Table 2 and the corresponding variables are in Fig. 2. $R^2$ values show correlation between each CTM and its own 5-parameter model. GEOS-Chem simulations use either MERRA or GEOS-5 meteorology. All lifetimes are smoothed with a 12-month running mean.
Fig. 2. Climate and emission variables controlling the interannual variation of $\tau_{\text{CH}_4 \times \text{OH}}$ in CTMs. Emissions are global totals, while other climate variables are averaged over 40°S–40°N, where 80% of methane oxidation occurs. Colors indicate which inputs are used by each CTM.

Fig. 3. Sensitivity of $\tau_{\text{CH}_4 \times \text{OH}}$ to biomass burning emissions, $E$, in the UCI CTM. Biomass burning CO emissions and the CO/NO emission ratio from the GFED3 inventory are also shown. Peak emissions and CO/NO ratio occur during El Niño events, due to tropical peat fires.
Fig. 4. Interannual variability of \( \tau_{\text{CH}_4 \times \text{OH}} \) in CTMs and observed methyl chloroform (MCF) decay rate. Observations are derived from atmospheric MCF abundances at NOAA and AGAGE surface stations (Montzka et al., 2000; Prinn et al., 2005), with an uncertainty (shaded) given by the 16th to 84th percentile range (±1σ) of decay rates across stations within each network, and adjusted by the tropospheric OH fraction of total MCF loss. All data are shown as anomalies relative to their own 2004–2010 mean (2004–2008 for NOAA data). Models, observations, and uncertainties are smoothed with a 12-month running average. Note anomalies in \( \tau_{\text{CH}_4 \times \text{OH}} \) and decay rate have opposite sign.

Fig. 5. Recent historical \( \tau_{\text{CH}_4 \times \text{OH}} \) variation (left) and its component causes (right). Lifetime reconstructed in this work from Eq. (1), with components shown at right. Shaded region shows ±σ uncertainty propagated from parameter ranges in Table 2, but not including possible errors in reanalysis, ozone, or emission inputs. All data are anomalies with respect to their 2000–2005 means. The anthropogenic (anthro) NO\(_x\) component combines the separate effects of land, ship, and aircraft emissions. See Fig. S6 for uncertainties in each component.
Fig. 6. Contribution of emissions and chemistry-climate interactions to changes in $\tau_{\text{CH}_4,\text{OH}}$ from 1980 to 2005 (top) and from 2010 to 2100 (bottom). Components and their uncertainties are derived from parameters in Table 2 and forcing variables in Table 3. Uncertainties (vertical bars) are standard deviations from $10^5$ monte carlo integrations. Note the different vertical scales.

Fig. 7. Projected future methane abundance (left) and total lifetime (right) for RCP 8.5. Projected uncertainty (shaded) is the standard deviation from $10^5$ monte carlo integrations, accounting for uncertainty in the present-day budget, emissions, and climate-chemistry effects on $\tau_{\text{CH}_4,\text{OH}}$. Our projections are compared to MAGICC model (Meinshausen et al., 2011a) and the IPCC TAR formula (Prather et al., 2001).